Keratin Protein Modified Anatase TiO₂ Nanoparticle – Characterizations of Protein Functionalized Surface

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Abstract

Anatase titanium dioxide (TiO₂) nanoparticle surface is modified with Keratin, a protein from swine wool. The protein structure and the modified surface of the TiO₂ nanoparticle is investigated. The structure and bonding of the adsorbed protein on TiO_2 surface were investigated by Fourier-transform infrared spectroscopy (FT-IR). The surface morphology of hybrid nanocomposite was observed by Field Emission Scanning Electron Microscopy (FESEM). Chemical bonding between protein and titanium dioxide was further investigated by X-ray Photoelectron Spectroscopy (XPS). Electrochemical cyclic voltammetry (CV) was used to characterize the electrochemical behaviour the protein modified TiO₂. The protein structure conformation and chemical bonding interaction with TiO₂ surface critically depend on the protein concentration. The protein is likely adsorbed via the interactions between the methyl, carbonyl, and amines group on the side chains of random coil secondary structure with the anatase TiO_2 surface. The protein has strong bonding to TiO_2 surface as evidently observed XPS for an environment bonding change in binding energy and component at OIs Ti2p and NIs region. The electrochemical behaviour of keratin protein modulated anatase TiO₂ hybrid material shows significant influence by protein structure and/or protein concentration at the surface.

Keywords: Keratin, Titanium dioxide, Anatase, Hybrid materials

1. Introduction

Organic-inorganic conjugated nanohybrid presents new superior unique properties for many applications such as drug delivery, cancer imaging, catalyst, energy storage, implant biosensors and bioelectronics [1-4]. However, the interface phenomena on protein structure, surface structure, adsorbed bonding interaction, synthesis, application and characterization of organic protein functionalization of metal oxide nanoparticle surface still require further study [5]. Titanium dioxide (TiO₂) is one of the most important semiconductor materials that attracts great interest in many applications due to unique band gap property, resource abundance, chemical and physical stability. In large volume consumption, TiO_2 has been widely used as pigment in paint, cosmetic and food industries. For high value applications such as dyesensitized solar cell, photocatalyst, photoelectrochemical, sensors, biomedical, bioelectronics, battery and supercapacitor, TiO₂ has been explored and utilized in form of modified structure and surface functional material [6-8]. TiO₂ conjugated with protein biomolecules surprisingly exhibits new functional challenges in many applications [8]. Proteins can interact with TiO₂ surface by different ways such as low interaction of ionic bonding, hydrogen bonding, and strong interaction of covalent bonding [5, 9] from cleavage bonding of 3D structure protein, such as cysteine disulfide bonds, hydrogen bonds and ionic bonds on the side protein chain. The bonding interaction between a protein and TiO₂ surface highly depends on many factors including protein type, protein structure and conformation, TiO₂ surface lattice and morphology structure, TiO_2 surface area, protein concentration, temperature and pH. Keratin is a low cost protein waste from feathers and food industry with a high content of cysteine

disulfide bond, glycine, and serine. These functional groups are suitable for interactions with TiO_2 surface to prepare TiO_2 protein conjugate hybrid material for the previously mention applications.

In this work, keratin protein adsorption bonding and conformation structure denature at anatase TiO_2 surface were investigated by XPS and IR at the amide and methyl group region. The electrochemical performance on cyclic voltammetry as capacitance electrode is also reported.

2. Experimental

2.1 Chemical and Materials

Sodium hydroxide (NaOH) derived from Univar, Ajax Finechem Pty Ltd was used as a washing solvent. The commercial anatase TiO_2 was purchased from UNILAB, Ajax Finechem Pty Ltd. The swine wool waste was collected from a community enterprise slaughterhouse of grill swine in Samphran district, Nakhon-Pathom province.

2.2. Methods

The Swine wool waste was washed by stirring in water at 60°C for 2 h, stirring in 0.1M NaOH at 60°C for 6 h, and washed by DI water until the pH of the solution was close to neutral. Keratin protein solution was prepared from 2.5 g and 10.0 g of swine wool in 0.5 M NaOH (500 mL). The starting 1.0 g anatase TiO₂ was mixed with 50 mL of keratin protein solution under stirring at room temperature for 2 h. The sediment sample was washed several times by DI water and dried at 60°C for 12 h. The prepared anatase TiO₂ samples from 2.5 g and 10.0 g of swine wool solution are named as A-2.5Kr and A-10Kr, respectively.

2.3 Characterizations

Surface morphologies of the prepared samples were observed by FESEM (HITACHI-S47000). The IR characteristics of the protein conformation and structures were investigated by FTIR (IRTracer-100 Shimadzu) with 40 scans. Chemical surface bonding environment of the adsorbed keratin protein on TiO_2 surface was further studied using XPS (AXIS Ultra DLD, Kratos Analytical, Shimadzu). Electrochemical performance on cyclic voltammetry was conducted by working planar carbon paste electrode 4×5 mm² VS Ag/AgCl in 1M KCl electrolyte. Approximately 3 mg of active sample was placed at working electrode without using any additive or binder. The cyclic curve data was collected by NOVA 2.1 software with Autolab PGSTAT302N.

3. Results and discussion

Fig. 1 shows FESEM images of (a) A-2.5Kr and (c) A-10Kr samples at low magnification to observe overall sample particle morphologies. High magnified images illustrated in (b) A-2.5Kr and (d) A-10Kr reveal surface morphologies modulated by keratin protein. The clean surface was observed indicating that protein structure could be in the primary and/or secondary structure conformation because the 3D tertiary and quaternary structure should show the morphological structure at least in the regime of 5 nm [10, 11].

FTIR analysis of A-2.5Kr and A-10Kr samples compared with the unmodified anatase TiO_2 were carried out and the corresponding overview spectra from 500 to 4500 cm⁻¹ are exhibited in Fig. 2 (a). FTIR spectra of all samples show prominent main peak of Ti-O lattice of TiO₂ for Ti-O bending mode at 500 – 600 cm⁻¹ corresponding to main composition of anatase TiO₂ [12].



Fig. 1. Low and high magnitude FESEM image of (a,b) A-2.5Kr and (c,d) A-10Kr samples.

For the protein characteristic pattern, a low transmittance intensity pattern is assigned to thin layer keratin protein binding at TiO_2 surface. It also implies the existence of the protein at the surfaces of TiO_2 particles. Methyl and amide regions are observed as an enlarged IR patterns as shown in Fig. 2 (b) and (c). The IR pattern of the amide group implies the secondary conformation structure of the protein. Vibration energy at around 2966, 2972 and 2853 cm⁻¹ are assigned to C-H of CH₃ asymmetrical stretching, C-H of $-CH_2$ - asymmetrical stretching and CH₂ symmetrical stretching, respectively [13]. A slightly different methyl peak centroid between A-2.5Kr and A-10Kr is probably due to different protein structure conformations. The difference C-H conformation of adsorbed protein structure may be due to different alkali ratio and/or different protein concentrations which may lead to different interaction side between C-H groups of adsorbed protein with TiO₂ surface.

Amide I and II which corresponds to the peaks around 1652 cm⁻¹ of A-2.5Kr and A-10Kr and 1492 cm⁻¹ of A-2.5Kr [14, 15] are clearly observed. The vibration peak at 1432 cm⁻¹ which deviates between amide II and III indicates the existence of N-N and/or C-N bond conformation due to bonding with TiO₂ surface. The amide I peaks present peak centroid and shape which resemble secondary protein of random coil structure and probably combines with small degree of β -sheet structure. Both keratin protein characteristics of methyl and amide compared with bare anatase TiO₂ particle proves that the keratin protein could be adsorbed on TiO₂ surface in both A-2.5Kr and A-10Kr samples. The peak at 1052 cm⁻¹ is attributed to stretching mode of sulfonate S-O or S=O that may arise from disulfide bond cleavage and interaction with TiO₂ surface.



Fig. 2. ATR-FTIR spectra A-10Kr, A-2.5Kr samples and the starting anatase TiO₂.

Survey scan XPS spectra of A-10Kr, A-2.5Kr samples and the anatase TiO₂ as presented in Fig. 3 further reveal interactions between the protein and TiO₂ surface. The survey XPS spectra are able to determine the main elements O and Ti corresponding to the main composition of anatase TiO₂ without significant impurity element. The high resolution at C1s, O1s, Ti2p, and N1s were examined for TiO₂ and biomolecule bonds environment. Curve fitted XPS spectra of A-10Kr and A-2.5Kr samples at C1s were performed according to carbon bond binding energy of proteins and adsorbed carbon impurity as shown in Fig. 4 (a). The main peak positioned at 284.8 eV ascribes to C-C and C-H bonds. The higher binding energy peaks at 288.6 eV may correspond to the C=O, N-C=O groups and at around 285.9-286.7 eV may correspond to the C-N, C-O-H groups of the protein in the A-10Kr and A-2.5Kr samples. However, the peaks at 285.9 to 288.6 eV also represents in bare anatase TiO₂ sample due to adsorbed carbon impurity.



Fig. 3. Survey scan XPS spectra of A-10Kr, A-2.5Kr samples and the starting anatase TiO₂.



Fig. 4. High resolution XPS spectra of A-10Kr, A-2.5Kr samples and the starting anatase TiO_2 at the region of (a) C1s and (b) O1s.

The fitting curve of O1s spectra also indicates the main composition of Ti-O-Ti oxygen in TiO₂ lattice at around 529.6 eV of bare anatase TiO₂ as exhibited in Fig. 4 (b). The O1s oxygen in the lattice of A-2.5Kr shifts toward lower binding energy accounting for oxygen around oxygen vacancy site of surface defect which may be induced by alkali solution. The shift to 529.7 of A-10Kr indicates lattice surface oxygen bonding with protein [11,12]. Bonding environment of keratin protein, such as O=C-N, C=O/O-C-H and C-O/H₂O typically appear around 530.9, 531.8 and 533.3 eV which are surprisingly noticed in bare anatase TiO₂.

As depicted in Fig. 5 (a), Ti2p spin orbit energy splitting component of Ti2 $p_{1/2}$ and Ti2 $p_{3/2}$ for TiO₂ at around 5.7 eV for A-10Kr. A-2.5Kr samples and the starting anatase TiO₂ is observed in Ti2 $p_{3/2}$ showing binding energy peaks around 458.4 eV [17]. The main Ti2 $p_{3/2}$ peak shows only main composition of Ti⁴⁺ in TiO₂ lattice environment that is in good agreement with other results while this value in A-2.5Kr shifts to 458.3 eV due to surface oxygen vacancy that is in accordance with the result from O*Is* spectra.

There is no significant signal of sulfur since a typical sulfur signal in protein was disappeared during dissolution in alkali solution. However, the N1s photoelectron appears in A-10Kr and A-2.5Kr samples showing the relevant evidence of amino protein $-NH_2$, $-NH_-$ and -N= at surface particle at binding energy of 400.0 eV [16,18,19]. The electrochemical capacitance behavior of A-10Kr and A-2.5Kr samples were performed by cyclic voltammetry and the curves of different scan rates are represented in Fig. 6 (a) and (b) for capacitance retention profiles. The CV shapes and current density property of A-10Kr and A-2.5Kr samples are compared in Fig.7 for 100 mV/s indicating that CV profile of A-2.5Kr shows redox peaks when compared with A-10Kr.



Fig. 5. High resolutions XPS spectra of A-10Kr, A-2.5Kr samples and the starting anatase TiO_2 in the region of (a) Ti2p and (b) N*Is*.



Fig. 6. Cyclic voltammmogram curves of (a) A-10Kr and (b) A-2.5Kr samples with different scan rates soaked in 1.0 M KCl electrolyte.



Fig. 7. Cyclic voltammogram curves of A-10Kr and A-2.5Kr samples measured at a scan rate of 100 mV/s in 1.0 M KCl electrolyte solution.

4. Conclusion

Keratin protein was conjugated to TiO_2 surface in the form of secondary random coil conformation structure by bonding via methyl, carbonyl and amine groups. Keratin protein concentration and alkali are considered as the major parameters that have strong effect on adsorbed protein structure conformation behavior and corresponding electrochemical property. TiO_2 nanohybrid material showed prominent property with feasibility for advance supercapacitor and bioelectronics technology.

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